

Name _____

Answer Key: Introduction to Atoms

1. Atoms are the fundamental building blocks of matter, and they are important because they make up everything around us.

2. An atom consists of a nucleus (containing protons and neutrons) and electrons in energy levels or electron shells.

3. Protons have a positive charge, and neutrons have no charge.

4. Electrons are involved in chemical reactions and bonding between atoms.

5. False